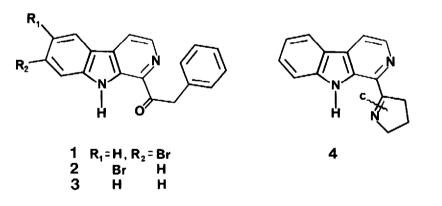
SHORT, EFFICIENT SYNTHESES OF THE ANTIBIOTIC EUDISTOMINS I AND T

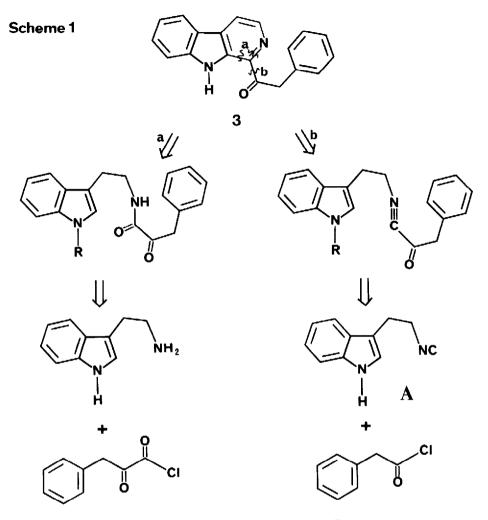
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Summary - Simple, concise syntheses of eudistomins I and T, β -carboline antibiotics from the tunicate <u>Eudistoma olivaceum</u>, have been achieved. The route utilized is an attractive alternative to traditional β -carboline syntheses and is amenable to preparation of a wide range of analogs.

Not long ago, we reported the isolation of seven B-carbolines from the Bermudian tunicate <u>Eudistoma olivaceum</u>, among them the unprecedented eudistomins R, S and T (1-3).¹ The very small quantities of these metabolites available from the ascidian precluded any biological testing. Reports of antimicrobial^{2,3} and antiviral³ activity in the other eudistomins¹⁻³ stimulated us to synthesize eudistomin T for pharmacological and agrochemical testing. Herein we describe concise synthetic approaches to eudistomins T, 3, and I, 4.



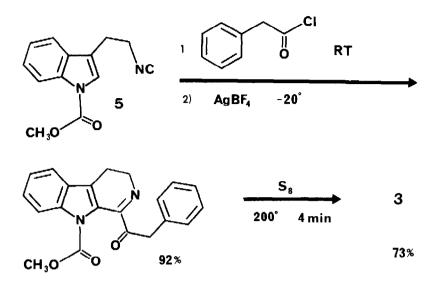
A prospective synthetic approach employing the traditional Bischler-Napieralski cyclization⁴ (bond disconnection "a", Scheme 1) would involve condensation of tryptamine with a phenyl pyruvyl halide, followed by cyclization and aromatization. Predicted⁵ and realized⁶ difficulties in condensing tryptamine with a variety of protected or masked phenyl pyruvic acid halides led us to consider an alternative approach, characterized by bond dissection "b" in Scheme



1. This approach, developed by the Livinghouse $group^7$, hinges upon the silver ion mediated cyclization of α -ketoimidoyl chlorides, which can be prepared by acylation of the isonitrile **A** by an acyl chloride. The dihydro β -carbolines thus obtained would then be dehydrogenated and deprotected to afford the target compounds.

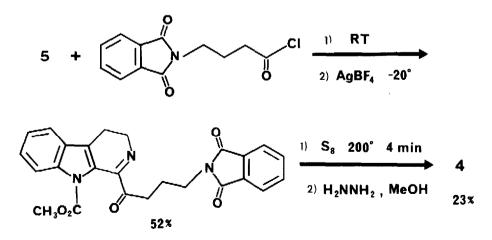
For the synthesis of eudistomin T (3), the N-protected indole isonitrile 5^{75} was acylated with phenylacetyl chloride; the reaction mixture was exposed to silver tetrafluoroborate to afford the dihydro-B-carboline 6 in 92% yield. A number of common dehydrogenation methods (Pd/C⁸, DDQ⁹, and ϕ_3 PRhCl¹⁰) were explored, but mediocre yields and complex purification problems resulted in each case. The most expedient route proved to be treatment of 6 with elemental sulfur at 200° for 4 minutes¹¹; both dehydrogenation and deprotection were accomplished in one step, providing eudistomin T (3)¹² in 73% yield.

Preparation of eudistomin I (4) by this method would require assembly of the



pyrrolidine ring subsequent to the acylation/cyclization step and protection during the aromatization. Dissection of the imine bond ("c" in 4) suggested that a suitably protected γ -aminobutyric acid, condensed with the isonitrile 5, could lead to 4. Treatment of the isonitrile 5 with 4-phthalimidobutyryl chloride, followed by silver tetrafluoroborate gave the dihydro- β -carboline 7 in 52% yield. Exposure of the intermediate 7 to elemental sulfur at 200° for 4 minutes provided 8; hydrolysis of 8 with methanolic hydrazine¹³ resulted in cleavage of the protecting groups and cyclization to eudistomin I, 4,¹⁴ in 23% yield.

Preliminary biological testing of eudistomin T revealed antimicrobial activity (inhibition of <u>Aspergillus terreus</u>, <u>Staphylococcus aureus</u>, <u>Bacillus cereus</u> and <u>Corynebacterium michiganense</u>), but no phytotoxicity toward johnsongrass or



3608

activity against the tobacco hornworm. A report on the synthesis of a series of analogs of 3 and their pharmacological and agrochemical profiles will be presented elsewhere.

The simplicity and versatility of this approach to ß-carbolines complements the Bischler-Napieralski reaction and should find numerous applications in the synthesis of members of this important class of pharmacologically active compounds.

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References

- 1. K.F. Kinzer and J.H. Cardellina II, <u>Tetrahedron</u> Lett., 28, 925 (1987).
- J.H. Cardellina II, <u>Pure Appl. Chem.</u>, 58, 365 (1986).
- 3. a) J. Kobayashi, G.C. Harbour, J. Gilmore and K.L. Rinehart, Jr., <u>J. Am.</u> <u>Chem. Soc.</u>, 106, 1526 (1984); b) K.L. Rinehart, Jr., J. Kobayashi, G.C. Harbour, R.G. Hughes, Jr., S.A. Mizsak and T.A. Scahill, <u>J. Am. Chem. Soc.</u>, 106, 1524 (1984); c) K.L. Rinehart, Jr., J. Kobayashi, G.C. Harbour, J. Gilmore, M. Mascal, T.G. Holt, L.S. Shield and F. Lafargue, <u>J. Am. Chem.</u> <u>Soc.</u>, 109, 3378 (1987).
- 4. P.E. Eaton, G.R. Carlson and J.T. Lee, <u>J. Org.</u> Chem., 38, 4071 (1973).
- 5. T. Hudlicky, T.M. Kutchan, G. Shen, V.E. Sutliff and C.J. Coscia, <u>J. Org.</u> <u>Chem.</u>, **46**, 1738 (1981).
- 6. N.-K. Lee and J.H. Cardellina II, unpublished observations.
- 7. a) M. Westling and T. Livinghouse, <u>Tetrahedron Lett.</u>, 26, 5389 (1985); b)
 M. Westling, R. Smith and T. Livinghouse, <u>J. Org. Chem.</u>, 51, 1159 (1986).
- 8. H.O. House and R.W. Bashe, <u>J. Org. Chem.</u>, **32** 784 (1967).
- 9. R.T. Arnold, C. Collins and W. Zenk, <u>J. Am. Chem. Soc.</u>, **62**, 983 (1940).
- 10. J. Blum and S. Biger, Tetrahedron Lett., 1825 (1970).
- 11. E.B. Hershberg and L.F. Fieser, Org. Syn., 2, 423 (1943).
- 12. Eudistomin T(3): λ_{max} (CH₂Cl₂) 380 nm ($\varepsilon = 7000$), 280(56100), 227(24100); ν_{max} (CDCl₃): 3440, 3065, 3035, 2930, 1672, 1627, 1597 cm⁻¹; MS: m/z 286.1085 (calc'd for C₁₉H₁₄N₂O - 286.1106) 286(65), 256(39), 195(20), 167(100), 140(37), 91(36); ¹H-NMR(CDCl₃): 8 10.3(1H,s), 8.58(1H, d, J=5), 8.14(1H, d, J=5), 8.12(1H, d, J=8), 7.51(1H, t, J=8), 7.55(1H, d, J=8), 7.39(1H, t, J=8), 7.40 - 7.25(5H,m), 4.74(2H,s).
- 13. T. Sasaki, K. Minamato and J. Itoh, <u>J. Org. Chem.</u>, **43**, 2320 (1978).
- 14. Eudistomin I(4): $\lambda_{masc}(CH_2Cl_2)$ 368 nm(ε =5700) 282(11300), 227(14500); $\nu_{masc}(CDCl_3)$: 3370, 3065, 3035, 2930, 1630, 1607, 1435 cm⁻¹; MS: m/z 235.1117 (calc'd for $C_{15}H_{13}N_3$ - 235.1047) 235(100), 206(33), 193(44), 167(15); ¹H-NMR(CDCl_3): δ 10.9(1H, s), 8.49(1H, d, J=5), 8.12(1H, d, J=8), 7.99(1H, d, J=5), 7.55(1H, dd J=5.5, 5), 7.51(1H, d, J=5), 7.31(1H, dd, J=8, 5.5), 4.25(2H, m), 3.33(2H, m), 2.07(2H, m).

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